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# STUDIES OF 7β-[2-(AMINOARYL)ACETAMIDO]-CEPHALOSPORIN DERIVATIVES

## I. SYNTHESIS AND STRUCTURE-ACTIVITY RELATIONSHIPS IN THE AMINOPYRIDINE SERIES

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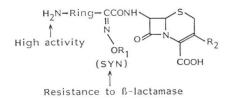
The synthesis and *in vitro* activity of  $7\beta$ -(2-aminopyridyl-2-alkoxyiminoacetamido)cephalosporins with various substituents at the 3-position are described. The effects of substitution pattern on the pyridine ring, oxime substituent and 3-substituent were studied as a function of the MIC values. Of these various kinds of derivatives,  $7\beta$ -[2-(2-aminopyridin-6-yl)-2-alkoxyiminoacetamido]cephalosporins exhibited significantly higher activity against most of microorganisms.

In the field of cephalosporins, the antibacterial activity strongly depends on the acyl moiety at the 7-position. Among a number of potent cephalosporins produced during the past twenty years, the aminothiazolyl oxyimino series such as ceftizoxime<sup>1</sup>, cefotaxime<sup>2,3</sup>, cefmenoxime<sup>4</sup> and ceftriaxone<sup>5</sup> in particular have not only excellent antibacterial properties but also marked resistance to  $\beta$ -lactamases. One of the common structural characteristics of these compounds is the amino function on the thiazole ring, and it would seem that this amino function is important for potent antibacterial activities.

Our effort was concentrated on elucidating the effect of the amino function on the aromatic ring and finding a new acyl moiety superior to the aminothiazole derivatives. The alternation of the heteroaromatic ring possessing an amino function has not been extensively studied in the past.

In a first approach to a new acyl moiety, we assumed that the structure shown in Scheme 1 may be necessary for high antibacterial activity and resistance to  $\beta$ -lactamase. In order to evaluate this concept, we chose pyridine as the ring and designed the synthesis of pyridyl side chain acids bearing the amino function and the oxyiminoacetic acid substituent at the various positions on the ring.

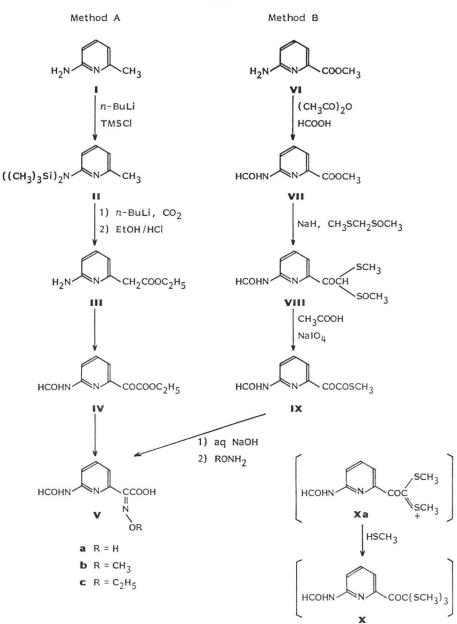
Scheme 1. Prerequisite to new acyl moieties.



We succeeded in synthesizing 2-amino-6-pyridyl-, 2-amino-5-pyridyl-, 2-amino-4-pyridyl- and 4amino-2-pyridylacetic acids possessing an oxime group in the  $\alpha$ -position, all of which were new compounds. The antibacterial activities of their cephalosporin derivatives were examined as a function of the MIC values.

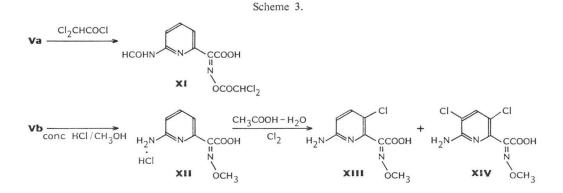
In this paper, we report the preparation of aminopyridyl side chain acids, the new antibiotics and the result of the structure-activity studies<sup>(0)</sup>.

Scheme 2.



### Chemistry

The side chain 2-(2-formamidopyridin-6-yl)-2-alkoxyiminoacetic acids were prepared by the two methods outlined in Scheme 2. According to Method A, aminopicoline (I) was converted to III *via* the disilylated compound (II), which was obtained by reaction with *n*-butyllithium and trimethylsilyl chloride (TMSCl). Attempts to carboxylate the *N*-acyl, phthaloyl or monosilylated derivatives of I were not successful. III was protected by formylation with a mixture of formic acid and acetic anhydride and oxidized with SeO<sub>2</sub> to afford the keto ester (IV), which was hydrolyzed with sodium hydroxide in 80% aqueous ethanol and then converted to the desired compound (V) by reaction with an appro-



priate oxyamine RONH<sub>2</sub>.

In addition, we developed an improved method which avoids the use of large amounts of *n*-butyllithium for a large scale preparations. According to Method B, methyl 2-formamidopyridyl-6-carboxylate (VII), obtained by *N*-protection of VI, was transformed to VIII by reaction with sodium hydride and methyl methylthiomethyl sulfoxide according to the OGURA method<sup>7</sup>. VIII was converted to the corresponding ketothioester (IX) by heating with sodium periodate in glacial acetic acid. In this rearrangement, without sodium periodate, a significant amount of X was obtained. This product (X) might have been generated from the intermediate (Xa) through addition of the methanethiol arising from the reaction. To transform the orthothioformate (X) into the ketothioester (IX), the reagent (NaIO<sub>4</sub>) was throughly effective, without any unfavored side reaction. However other known oxidizing reagents such as hydrogen peroxide, periodic acid and *m*-chloroperbenzoic acid were not effective. Compound IX thus obtained was converted to the desired compound (V) by hydrolysis and subsequent reaction with an appropriate oxyamine RONH<sub>2</sub>.

2-(2-Formamidopyridin-5-yl)-2-methoxyiminoacetic acid, 2-(2-formamidopyridin-4-yl)-2-methoxyiminoacetic acid and 2-(4-formamidopyridin-2-yl)-2-methoxyiminoacetic acid were obtained in a manner similar to that of Method B. However in the case of these compounds, a mixture of acetic anhydride and formic acid (volume ratio: approximately 1: 10) was used to convert **VIII** to the ketothioester (**IX**), because these formyl groups were very easily lost by heating in acetic acid.

The hydroxyiminoacetic acid (Va) was converted to the corresponding dichloroacetoxyimino compound (XI) by treating with dichloroacetyl chloride in order to protect the hydroxyl group. The dichloroacetoxy group was smoothly removed by treating with aqueous NaHCO<sub>3</sub>.

Monochloro and dichloro aminopyridine derivatives were prepared by chlorination of the *N*-deprotected compound (XII) of V (Scheme 3).

The cephalosporins were prepared either by coupling the 2-formamidopyridyl-2-alkoxyiminoacetic acid with  $7\beta$ -aminoceph-3-em-4-carboxylic acids possessing various substituents at the 3-position or by displacing the 3-acetoxy group with thiol compounds after acylation of  $7\beta$ -aminocephalosporanic acid. The acylated cephalosporins were deformylated by treatment with concentrated hydrochloric acid in methanol.

## Antibacterial Activity

Table 1 shows the MICs of aminopyridylcephalosporins possessing a methoxyimino moiety at the  $\alpha$ -position of the side chain acid. The 3-position of the cephalosporin nucleus is methyltetrazolylthio-

		N	сн <sub>3</sub>	соон	L N CH <sub>3</sub>			
Compound No.	R	S. aureus 209p JC-1	E. coli NIHJ JC-2	P. vulgaris IAM- 1025	P. aeruginosa NCTC- 10490	P. aeruginosa IAM-1095	S. marce- scens 35	E. cloacae 60
1	H <sub>2</sub> N	1.56	0.39	0.025	1.56	25	3.13	3.13
2	H <sub>2</sub> N N	1.56	3.13	0.39	12.5	400	>100	>100
3	H <sub>2</sub> N N	1.56	1.56	3.13	50	>800	>100	50
4	NH2	12.5	0.78	0.39	100	800	50	3.13
5	H <sub>2</sub> N N	6.25	25	0.78	50	800	>100	100
6	H <sub>2</sub> N CI	3.13	>100	50	400	>800	>100	>100

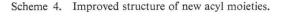
Table 1. Antibacterial activity (MIC  $\mu$ g/ml) of aminopyridyl cephalosporins.

R-CCONH S

methyl group which is typical in this field<sup>4,5)</sup>. In the 2-aminopyridine series (compounds 1, 2 and 3), the activity of compound 1 is significantly higher than the other two (2 and 3) against all organisms except *Staphylococcus aureus*. On the other hand, a shift of the amino function in the 4-aminopyridin-2-yl derivative (4) showed con-

6.25

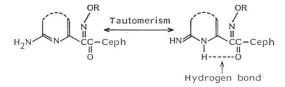
Ceftizoxime



25

1.56

6.25



siderable activity, but which was still lower than that of 1, especially against *Pseudomonas aeruginosa*.

0.025

0.025

0.39

These results suggest that the positions of the amino function and oxyiminoacetic moiety were very important for good antibacterial activities. Therefore we assumed that the partial structure shown in Scheme 4 might be necessary for good activities. Interestingly the partial structure is consist with the aminothiazolyloxyiminocephalosporins.

The successful synthesis of 2-aminopyridin-6-ylcephalosporin derivatives which exhibited potency comparable to that of ceftizoxime prompted us to make further chemical modifications.

The monochlorinated derivative (5) had poor activity against Gram-negative bacteria except *Proteus vulgaris* and the dichlorinated compound (6) was generally inactive against Gram-negative bacteria but exhibited higher activity than compound 5 against *S. aureus*.

Exchanging the alkoxyimino part led to no significant change of activity except that the hydroxyimino derivative (7) improved the activity against *S. aureus* and *E. coli*, but poorer activity against *P*.

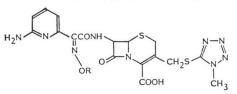
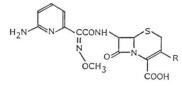


Table 2.	Antibacterial	activity (MIC	ug/ml)	of 2-aminopy	vridin-6-vl	cephalosporins.

Ρ. Ρ. E. coli S. marce-P. vulgaris aeruginosa IAM-1025 NCTC- IAM-1095 Compound S. aureus E. cloacae R NIHJ scens No. 209p JC-1 60 IAM-1095 JC-2 35 10490 7 Η 0.39 0.2 0.2 6.25 100 25 6.25 8  $C_2H_5$ 3.13 1.56 0.1 1.56 12.5 12.5 6.25 9 n-Pr 1.56 6.25 0.1 1.56 25 12.5 3.13 10 iso-Pr 3.13 6.25 0.2 25 6.25 1.56 3.13 11 n-Bu 0.78 6.25 0.78 1.56 12.5 12.5 3.13 12 iso-Bu 1.56 12.5 0.39 1.56 25 25 12.5 13  $CH_2CH = CH_2$ 3.13 3.13 0.05 1.56 12.5 6.25 3.13 1.56 3.13 14  $CH_2C \equiv CH$ 0.05 1.56 12.5 12.5 3.13 15 CH<sub>2</sub>CF<sub>3</sub> 3.13 6.25 0.1 1.56 25 25 6.25

Table 3. Antibacterial activity (MIC µg/ml) of 2-aminopyridin-6-ylcephalosporins.



Com- pound No.	R	S. aureus 209p JC-1	E. coli NIHJ JC-2	P. vulgaris IAM-1025	P. aeruginosa NCTC- 10490	P. aeruginosa IAM-1095	S. marce- scens 35	E. cloacae 60
16	CH <sub>2</sub> OCOCH <sub>3</sub>	3.13	0.39	0.025	1.56	25	25	12.5
17	$CH_2OCONH_2$	3.13	0.78	0.05	3.13	100	50	50
18	H <sub>2</sub> CS K <sub>S</sub>	1.56	0.78	0.025	1.56	12.5	12.5	3.13
19	H <sub>2</sub> CS K <sub>S</sub> CH <sub>2</sub> NH <sub>2</sub>	1.56	0.78	0.10	6.25	50	12.5	3.13
20	H <sub>2</sub> CS H <sub>2</sub> COOH	50	0.78	0.025	1.56	25	6.25	12.5
21	Н	12.5	0.20	0.025	6.25	400	12.5	12.5
22	$\mathbf{CH}_{3}$	100	25	1.56	100	>800	>100	>100
23	Cl	3.13	0.78	0.10	25	400	>100	>100

aeruginosa (Table 2).

Table 3 shows the activities of 2-aminopyridin-6-ylcephalosporins with various substituents at the 3-position. Heteroaromatic thiomethyl derivatives exhibited good activities, but none exceeded *N*-

methyltetrazolylthiomethyl compound except a slight improvement against P. aeruginosa in the thiadiazolylthiomethyl derivative (18).

As a result of our investigations, we have succeeded in finding a new acyl moiety which confers potent antibacterial activities, and have demonstrated experimentally that the partial structure shown in Scheme 4 is associated with potent antibacterial activities.

#### Experimental

Melting points were determined using a Thomas-Hoover capillary melting point apparatus and are uncorrected. IR spectra were taken on a Hitachi 260-10 spectrophotometer or Shimadzu IR-420 spectrophotometer. NMR spectra were recorded at 60 MHz on a JNM-PMX 60 NMR spectrometer and at 100 MHz on a Jeol-MH 100 NMR spectrometer using TMS as an internal standard. The following abbreviations are used: s singlet, d doublet, dd double doublet, t triplet, q quartet, m multiplet, ABq AB quartet, bs broad singlet. Organic solvents were dried over anhydrous MgSO<sub>4</sub> and all concentrations by evaporation were carried out *in vacuo*.

Determination of In Vitro Antibacterial Activity

All the *in vitro* antibacterial activities are given as MIC in  $\mu$ g/ml required to prevent growth of the bacterial culture. MIC's were determined by the agar dilution method using heart infusion agar (Difco) after incubation at 37°C for 20 hours with an inoculum size of about 10<sup> $\circ$ </sup> cfu/ml.

#### General Preparation of V

Method A:

1) 2-[*N*,*N*-Bis(trimethylsilyl)amino]-6-methylpyridine (II): A 15% *n*-hexane solution (636 g) of *n*-butyllithium was added to a solution of I (64.8 g) in tetrahydrofuran (500 ml) at -20 to  $-30^{\circ}$ C over 1 hour, and stirred at -8 to  $-10^{\circ}$ C for 30 minutes. To the solution was added trimethylsilylchloride (161.7 g) at -15 to  $-5^{\circ}$ C over 40 minutes, and the resultant solution was stirred at room temperature overnight. The solution was filtered through by a column packed with silica gel (180 g), washed with tetrahydrofuran and then the filtrate was concentrated. The residue was purified by fractional distillation to give II (127.6 g, 84.4%); bp 95~97^{\circ}C/5~6 mmHg.

2) Ethyl 2-(2-Aminopyridin-6-yl)acetate (III): A 15% *n*-hexane solution (338.6 g) of *n*-butyllithium was added dropwise to a solution of II (100 g) in anhydrous THF (300 ml) at -20 to  $-30^{\circ}$ C over 1 hour, then the solution was stirred at 20 to  $23^{\circ}$ C for 1 hour. The resultant solution was added in small portions to crushed dry ice (1 kg) with stirring, and stirring was continued until room temperature was reached. After removing tetrahydrofuran from the solution *in vacuo*, absolute ethanol (1 liter) was added to the residue. An anhydrous 30% ethanol solution (660 ml) of hydrochloric acid was added dropwise to the solution at -5 to  $-10^{\circ}$ C and further hydrogen chloride gas was bubbled at 0 to  $5^{\circ}$ C for 30 minutes, then the solution was stirred at  $10^{\circ}$ C overnight. After removing the ethanol, the residue was dissolved in water, and washed with ethyl acetate 3 times. The solution was adjusted to pH 7 to 8 with sodium bicarbonate and extracted with ethyl acetate. The extract was dried and concentrated to give the crude product (54 g, 75.0%). The product was purified by column chromatography on silica gel (1 kg) (eluant; EtOAc - C\_{6}H\_{6}, 3: 1) to give III (30.2 g, 41.9%); mp  $66 \sim 68^{\circ}$ C.

Anal Calcd for C<sub>9</sub>H<sub>12</sub>N<sub>2</sub>O<sub>2</sub>: C 59.98, H 6.71, N 15.55.

Found: C 59.77, H 6.50, N 15.21.

3) Ethyl 2-(2-Formamidopyridin-6-yl)glyoxylate (IV): Acetic anhydride (16.6 ml) and 98% formic acid (7.32 ml) were mixed at room temperature and stirred at 50 to 60°C for 30 minutes. The solution was added dropwise to a solution of III (26.5 g) in ethyl acetate (250 ml) at 20 to 23°C over 30 minutes, and stirred at the same temperature for 1 hour. Cold water was added to the resultant solution and the mixture was shaken thoroughly. The ethyl acetate layer was separated, washed with water, aqueous sodium bicarbonate and water in turn, dried and concentrated to give ethyl 2-(2-form-amidopyridin-6-yl)acetate (28 g, 91.5%); mp 35~38°C: IR (Nujol) 3250, 3100, 1738, 1690, 1580, 1460, 1305, 1277 cm<sup>-1</sup>. To a solution of the above compound (26 g) in dioxane (260 ml) was added selenium

dioxide (16.65 g) in small portions at 85 to 90°C over 1 hour and stirred at the same temperature for 1 hour. After cooling the resultant mixture, the dioxane layer was separated, concentrated and residue was dissolved in ethyl acetate. The solution was washed with water, and treated with activated charcoal and then concentrated to give IV (14.3 g, 51.5%); mp 124~126°C; IR (Nujol) 3220, 3100, 1737, 1720, 1690, 1273, 1233 cm<sup>-1</sup>.

4) 2-(2-Formamidopyridin-6-yl)-2-methoxyiminoacetic Acid (Vb, Z isomer):  $2 \times Sodium$  hydroxide solution in 80% aqueous ethanol (14.9 ml) was added to a solution of IV (6.0 g) in ethanol (180 ml) at room temperature and stirred at that temperature for 20 minutes. Methoxyamine hydrochloride (2.7 g) was added to the resultant solution, stirred at room temperature for 1.5 hours and then concentrated to a small volume. The precipitates were collected by filtration, washed with ethyl acetate and water, dissolved in methanol and then treated with activated charcoal. The solution was concentrated and then the precipitates were collected by filtration to give Vb (Z isomer) (3.6 g, 60.2%), which was recrystallized from EtOAc - Et<sub>4</sub>O; mp 170~171°C (dec).

Anal Calcd for  $C_{\theta}H_{\theta}N_{\theta}O_{4}$ : C 48.43, H 4.06, N 18.83.

Found: C 48.79, H 4.28, N 18.84.

The configuration of the oxyimino group of Vb and the following side chain acids must be Z form since in these cephalosporin derivatives, the NMR chemical shift of the amide proton at the C-7 position was observed at very low field as same as that of aminothiazolyl cephems<sup>1,9</sup>.

Method B:

1) Methyl 2-Formamido-6-pyridinecarboxylate (VII): A mixture of formic acid (559.3 g) and acetic anhydride (1,033.4 g) was stirred for 30 minutes at 40 to 50°C and thereto was added VI<sup>10</sup> (616 g) at 40°C, and then the mixture was stirred for 1 hour at 80°C. After the removal of the solvent from the reaction mixture, to the residue was added a mixture of benzene (2 liters) and *n*-hexane (6 liters). The precipitates were collected by filtration and then recrystallized from benzene (2 liters) to give VII (647.8 g, 88.8%); mp 134~136°C.

2) 2-Formamido-6-(2-methanesulfinyl-2-methylthioacetyl)pyridine (VIII): To a mixture of VII (435.7 g), methyl methylthiomethyl sulfoxide (300 g) and *N*,*N*-dimethylformamide (2.2 liters) was added portionwise 50% oil suspension sodium hydride (348 g) with stirring and ice cooling, and the mixture was stirred for 30 minutes at room temperature. To the reaction mixture was added methylene chloride (4.4 liters) with ice cooling, the precipitates were collected by filtration and then added to a mixture of methylene chloride (3 liters), ice (2 kg) and concentrated hydrochloric acid (730 ml). The mixture was adjusted to pH 7 with sodium bicarbonate and extracted with methylene chloride. The extract was dried, concentrated and crystallized from diethyl ether to give VIII (430 g, 65.3%); mp 125~128°C.

3) S-Methyl 2-(2-Formamidopyridin-6-yl)thioglyoxylate (IX): A mixture of VIII (424 g) and sodium periodate (100 g) in acetic acid (2.1 liters) was stirred for 30 minutes at 70°C. After the removal of acetic acid, to the residue were added water (5 liters) and sodium thiosulfate (116 g), and then the mixture was adjusted to pH 7 with sodium bicarbonate. The precipitates were collected by filtration, washed with water and then dried to give IX (246.4 g, 70.6%), which was recrystallized from EtOAc; mp 170~ 173°C.

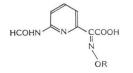
Anal Caled for C<sub>9</sub>H<sub>8</sub>N<sub>2</sub>O<sub>8</sub>S: C 48.21, H 3.59, N 12.49, S 14.30. Found: C 48.28, H 3.62, N 12.59, S 14.54.

In this reaction, the treatment of VIII in acetic acid without sodium periodate afforded IX in a low yield of 30.3% and 2-formamido-6-(2,2,2-trimethylthioacetyl)pyridine (X) in 48.7% yield which was separated by silica gel column chromatography, eluting with  $C_6H_6$  - EtOAc, 4:1. In addition, X was converted into IX in 74.8% yield by heating with sodium periodate in acetic acid. Compound X was crystallized from diisopropyl ether; mp 93~95°C; IR (Nujol) 1690, 1680, 1595, 1580, 1450, 1380, 1370, 1320, 1290, 1265 cm<sup>-1</sup>; NMR (DMSO- $d_6$ )  $\partial$  2.04 (9H, s), 6.8~7.1 (1H, m, ring proton), 7.4~7.9 (2H, m, ring protons), 10.4~10.7 (1H, m, NH).

Substrate		VII	VIII	IX	Vb
H2N COOCH3	Yield (%) mp (°C) IR $\nu_{C=0}^{N u j \circ 1}$ (cm <sup>-1</sup> )	88.1 134~136 1740, 1700	65.3 130~132 1710, 1690	70.6 163~165 1700, 1670	57.4 170~171 1745, 1680
H <sub>2</sub> N COOCH <sub>3</sub>	Yield (%) mp (°C) IR $\nu_{C=0}^{N ujo1}$ (cm <sup>-1</sup> )	90.6 218~220 1720, 1710	65.0 125~127 1710, 1660	69.5 152~154 1730, 1680	34.8 159~161 1735, 1665
H <sub>2</sub> N NH <sub>2</sub>	Yield (%) mp (°C) IR $\nu_{C=0}^{N ujo1}$ (cm <sup>-1</sup> )	94.2 190~197 1740, 1710	69.3 123~125 1700, 1690	49.0 165~167 1710, 1680	56.6 170~172 1710, 1640
N COOCH3	Yield (%) mp (°C) IR $\nu_{C=0}^{Nujo1}$ (cm <sup>-1</sup> )	68.7 185~186 1690, 1675	20.0 132~133 1700, 1680	65.0 145~148 1690, 1670	90.0 168~170 1725, 1650

Table 4. Yield, mp and IR data of regio-isomers of aminopyridines obtained by Method B.

Table 5. Yield, mp and IR data of 2-(2-formamidopyridin-6-yl)-2-alkoxyiminoacetic acids.



R	V:-14 (0/)	mm (°C dae)		IR (Nujol cm <sup>-1</sup> )	
K	Yield (%)	mp (°C dec) $-$	NH	СООН	СНО
Н	48.2	190~192	3120	1700	1665
$C_2H_5$	48.0	183~184	3220	1760	1680
<i>n</i> -Pr	45.6	155~156	3250	1740	1650
iso-Pr	34.7	140~142	3250	1755	1670
<i>n</i> -Bu	40.5	140~145	3300	1750	1670
iso-Bu	38.5	129~131	3150	1755	1670
$CH_2CH = CH_2$	50.6	153~155	3250	1750	1680
$CH_2C\equiv CH$	55.2	140	3250	1760	1670
$CH_2CF_3$	38.0	$145 \sim 150$	3250	1755	1685

4) 2-(2-Formamidopyridin-6-yl)-2-ethoxyiminoacetic Acid (Vc, Z isomer): A mixture of IX (4.5 g), methanol (20 ml) and 1 N aqueous sodium hydroxide (20 ml) was stirred for 50 minutes at room temperature to give a solution containing 2-(6-formamidopyridin-2-yl)glyoxylic acid. To the solution was added *O*-ethylhydroxylamine hydrochloride (2.2 g), and the mixture was stirred for 35 minutes at that temperature. The reaction mixture was adjusted to pH 7 with hydrochloric acid and the methanol was distilled off. The remaining aqueous mixture was washed with ethyl acetate, then layered with ethyl acetate and adjusted to pH 1 with 10% hydrochloric acid. The ethyl acetate layer was separated, washed with water, treated with activated charcoal and then concentrated to give Vc (Z isomer) (2.3 g, 48.0%), which was recrystallized from EtOAc - Et<sub>2</sub>O; mp 183~184°C (dec).

Anal Calcd for  $C_{10}H_{11}N_{8}O_{4}$ : C 50.63, H 4.67, N 17.72.

Found: C 50.21, H 4.89, N 17.35.

2-Aminopyridin-5-yl, 2-aminopyridin-4-yl, 4-aminopyridin-2-yl derivatives were obtained respectively from 2-amino-5-pyridinecarboxylate<sup>11)</sup>, 2-amino-4-pyridinecarboxylate<sup>12)</sup>, 4-amino-2-pyridinecarboxylate<sup>13)</sup> according to Method B. The properties of these compounds are listed in Table 4. 2-(2-Formamidopyridin-6-yl)acetic acids derivatives bearing various oxime substituents which were obtained by the reaction with the corresponding alkoxyamine are listed in Table 5.

#### 2-(2-Formamidopyridin-6-yl)-2-dichloroacetoxyiminoacetic Acid (XI, Z isomer)

A mixture of Va (3.6 g), dichloroacetyl chloride (7.6 g) and methylene chloride (100 ml) was stirred at room temperature for 5 hours. The precipitates were collected by filtration, washed with diethyl ether and dried to give XI (4.6 g, 83.6%); mp 88~90°C; IR (Nujol) 1800, 1720, 1620 cm<sup>-1</sup>.

#### Preparation of the Monochloro and Dichloro Derivatives (XIII, XIV, Z isomer)

A mixture of Vb (Z isomer) (5.0 g) and concentrated hydrochloric acid (2.3 ml) in methanol (50 ml) was stirred for 40 minutes at room temperature. After the removal of methanol from the reaction mixture, the residue was pulverized in diethyl ether, collected by filtration to give 2-(2-aminopyridin-6-yl)-2-methoxyiminoacetic acid hydrochloride (XII) (Z isomer) (5.2 g, ~100%) as a pale brown powder; NMR (DMSO- $d_6$ +D<sub>2</sub>O)  $\hat{\sigma}$  4.10 (3H, s), 6.84 (1H, d, J=7 Hz), 7.23 (1H, d, J=10 Hz), 7.99 (1H, dd, J=7 Hz, J=10 Hz).

To a mixture of XII (8.0 g), acetic acid (350 ml) and water (10 ml) was introduced chlorine gas for 1 hour at 8°C. After the removal of excess chlorine gas by bubbling air into the reaction mixture, the solvent was distilled off. The residue was pulverized in diethyl ether and collected by filtration. After the addition of water and ethyl acetate to the resultant powder (9.8 g), the aqueous layer was separated and washed with ethyl acetate twice and then the water was distilled off. The remaining water in the residue was azeotropically removed with benzene twice to yield a brownish powder, which was dried in a desiccator to give 2-(2-amino-5-chloropyridin-6-yl)-2-methoxyiminoacetic acid (XIII) (Z isomer) (3.3 g, 41.3 %); mp 142~145°C; NMR (DMSO- $d_6$ +D<sub>2</sub>O)  $\delta$  3.81 (3H, s), 6.50 (1H, d, J=9 Hz), 7.48 (1H, d, J=9 Hz).

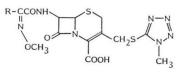
The remaining ethyl acetate layer was dried and concentrated and the residue was washed with diethyl ether to give 2-(2-amino-3,5-dichloropyridin-6-yl)-2-methoxyiminoacetic acid (XIV, Z isomer) (2.4 g, 26.3 %); mp 139 ~ 144°C; NMR (DMSO- $d_6 + D_2O$ )  $\delta$  3.96 (3H, s), 7.83 (1H, s).

General Procedure for the Acylation of 7<sup>β</sup>-Aminoceph-3-em-4-carboxylic Acids

A) Synthesis of Cephalosporins  $1 \sim 18$ , 22 (Z isomer): (1) Phosphoryl chloride (6.5 mmol) was added to N,N-dimethylformamide (5 ml) and stirred at 40°C for 30 minutes. To the solution was added a solution of 2-(2-formamidopyridin-6-yl)-2-alkoxyiminoacetic acid (V) (5 mmol) in N,N-dimethylformamide (5 ml) at  $-15^{\circ}$ C, and stirred at -10 to  $-8^{\circ}$ C for 50 minutes [solution A]. Separately, the 7-aminocephem derivative (5 mmol) and trimethylsilylacetamide (7.2 g) were dissolved in methylene chloride (20 ml) at 40°C and cooled. To the cool solution was added the above solution A at -20 to  $-15^{\circ}$ C and stirred at that temperature for 40 minutes. The resultant solution was poured into a solution of saturated aqueous sodium bicarbonate (30 ml) and water (40 ml) with ice cooling. The aqueous layer was separated, washed with ethyl acetate, and then ethyl acetate (50 ml) was added to the aqueous layer and adjusted to pH 3 with  $10^{\circ}$  hydrochloric acid. The organic layer was separated and extracted with ethyl acetate twice. The extracts were combined, washed with water and concentrated to a small volume. The resulting precipitates were collected by filtration, washed with ethyl acetate and dried, to give the acylated compounds in  $40 \sim 70^{\circ}$  yield.

(2) Conc hydrochloric acid (0.4 ml) was added to a solution of the above product (Z isomer, 2 mmol) in methanol (15 ml) and stirred at room temperature for 40 minutes. After the methanol removed, water (100 ml) was added to the residue, which was then dissolved by adding 10% hydrochloric acid. After some insoluble material was filtered off, the filtrate was adjusted to pH 3 with aqueous sodium bicarbonate and then submitted to column chromatography on macroporous, non-ionic adsorption resin (Diaion HP-20), with an eluant of aqueous methanol. The eluate was lyophilized to give the desired cephalosporins ( $40 \sim 65\%$ ).

B) Synthesis of Cephalosporins **19**, **20** (*Z* isomer): A solution of 7-[2-(2-formamidopyridin-6yl)-2-methoxyiminoacetamido]cephalosporanic acid (*Z* isomer) (5 mmol) and disodium 2-(5-mercapto-1*H*-tetrazol-1-yl)acetate (6 mmol) in water (40 ml) was adjusted to pH 7 with sodium bicarbonate, and stirred at 65°C for 6 hours at pH 7 to 7.4. The resultant solution was washed with ethyl acetate, adjusted to pH 2.5 with 10% hydrochloric acid and stirred. The precipitates were collected by filtration, washed with water and diethyl ether to give 7-[2-(2-formamidopyridin-6-yl)-2-methoxyiminoacetamido]-3-[(1-carboxymethyl-1*H*-tetrazol-5-yl)thiomethyl]-3-cephem-4-carboxylic acid (*Z* isomer) (1.3 g, 44.0%); Table 6. IR and <sup>1</sup>H NMR data of aminopyridyl cephalosporins.

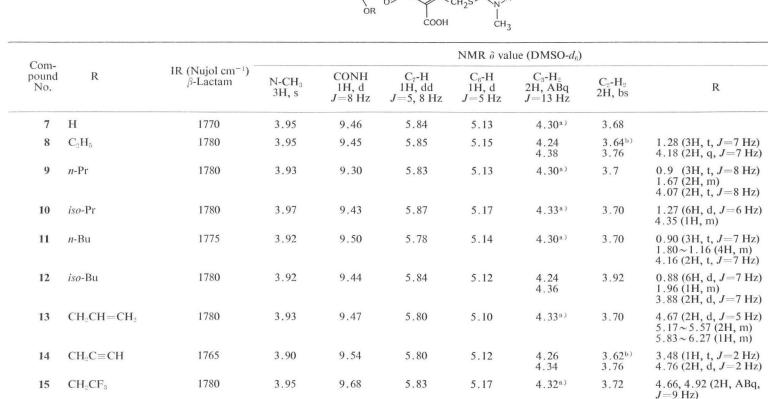


		NMR $\delta$ value (DMSO- $d_{\delta}$ )							
R	IR (Nujol cm <sup>-1</sup> ) $\beta$ -Lactam	N-CH <sub>3</sub> 3H, s	O-CH <sub>3</sub> 3H, s	CONH 1H, d J=8 Hz	$C_7-H$ 1H, dd $J=5, 8$ Hz	$C_6$ -H 1H, d J=5 Hz	$\begin{array}{c} C_3-H_2\\ 2H, ABq\\ J=13 \text{ Hz} \end{array}$	$C_2$ - $H_2$ 2H, bs	Ring proton
H2N	1780	3.75	3.78	9.52	5.85	5.17	4.30 4.37	3.73	6.53 (1H, d, <i>J</i> =8 Hz) 6.93 (1H, d, <i>J</i> =8 Hz) 7.48 (1H, t, <i>J</i> =8 Hz)
H <sub>2</sub> N N	1780	3.90	3.97	9.73	5.82	5.17	4.35 <sup>a</sup> )	3.75	6.57 (1H, d, <i>J</i> =9 Hz) 7.67 (1H, dd, <i>J</i> =2, 9 Hz) 8.03 (1H, d, <i>J</i> =2 Hz)
H <sub>2</sub> N N	1775	3.98	3.97	9.79	5.83	5.18	4.35 <sup>a)</sup>	3.73	6.67~6.80 (2H, m) 8.00 (1H, d, <i>J</i> =6 Hz)
	1778	3.77	3.97	9.52	5.80	5.12	4.32 <sup>a)</sup>	3.64	6.60 (1H, dd, <i>J</i> =2, 7 Hz) 6.97 (1H, d, <i>J</i> =2 Hz) 8.00 (1H, d, <i>J</i> =7 Hz)
H <sub>2</sub> N LNC	1790	3.94	4.09	9.70	5.80	5.14	4.21 4.37	3.70	6.97 (1H, d, <i>J</i> =10 Hz) 7.80 (1H, d, <i>J</i> =10 Hz)
H <sub>2</sub> N CI	1785	3.98	3.98	9.43	5.81	5.17	4.35 <sup>a)</sup>	3.77	7.87 (1H, s)
	$H_2N \xrightarrow{N} H_2N \xrightarrow{N} H_2N \xrightarrow{N} H_2N \xrightarrow{N} H_2$		R     β-Lactam     N-CH <sub>3</sub> 3H, s $H_2N$ 1780     3.75 $H_2N$ 1780     3.90 $H_2N$ 1775     3.98 $H_2N$ 1778     3.77 $H_2N$ 1778     3.77 $H_2N$ 1778     3.77 $H_2N$ 1778     3.94 $CI$ 1785     3.98	Kβ-LactamN-CH3 3H, sO-CH3 3H, s $H_2N$ 17803.753.78 $H_2N$ 17803.903.97 $H_2N$ 17753.983.97 $H_2N$ 17753.983.97 $H_2N$ 17783.773.97 $H_2N$ 17783.773.97 $H_2N$ 17783.944.09 $H_2N$ 17853.983.98	Rβ-LactamN-CH <sub>3</sub> 3H, sO-CH <sub>3</sub> 3H, sCONH IH, d J=8 Hz $H_2N$ 17803.753.789.52 $H_2N$ 17803.903.979.73 $H_2N$ 17753.983.979.79 $H_2N$ 17783.773.979.52 $H_2N$ 17783.773.979.52 $H_2N$ 17783.773.979.52 $H_2N$ 17853.983.989.43	RIR (Nujol cm^{-1}) $\beta$ -LactamN-CH3 $3H, s$ O-CH3 $3H, s$ CONH $IH, d$ $J=8 Hz$ C7-H $IH, dd$ $J=5, 8 Hz$ H2NI17803.753.789.525.85H2NI17803.903.979.735.82H2NI17753.983.979.795.83H2NI17753.983.979.795.83H2NI17753.983.979.525.80H2NI17783.773.979.525.80H2NI17783.773.979.525.80H2NI17783.773.979.525.80H2NI17783.944.099.705.80H2NI17803.983.989.435.81	RIR (Nujol cm^{-1}) $\beta$ -LactamN-CH3 $3H, s$ O-CH3 $3H, s$ CONH $1H, d$ $J=8 Hz$ C7-H $1H, dd$ $J=5, 8 Hz$ C6-H1 $IH, dd$ $J=5, 8 Hz$ C6-H1 $IH, dd$ $J=5 Hz$ H2N17803.753.789.525.855.17H2N17803.903.979.735.825.17H2N17753.983.979.795.835.18H2N17753.983.979.795.835.18H2N17783.773.979.525.805.12H2N17783.773.979.525.805.12H2N17803.944.099.705.805.14	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$\begin{array}{c c c c c c c c c c c c c c c c c c c $

a) bs.

Table 7. IR and <sup>1</sup>H NMR data of cephalosporins possessing various oxime parts.

H2N.



THE JOURNAL OF ANTIBIOTICS

MAY 1984

542

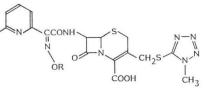


Table 8. IR and <sup>1</sup>H NMR data of cephalosporins with various substituents at the 3-position.

# H<sub>2</sub>N N CCONH S N OCH<sub>3</sub> COOH

					NMR $\delta$ value (DMSO- $d_6$ )					
Compound No.	R	IR (Nujol cm <sup>-1</sup> ) $\beta$ -Lactam	O-CH <sub>3</sub> 3H, s	CONH 1H, d J=8 Hz	$C_7-H$ 1H, dd $J=5, 8$ Hz	$C_{6}$ -H 1H, d J=5 Hz	$\begin{array}{c} C_2 \text{-} H_2 \\ \text{2H, bs} \end{array}$	$\begin{array}{c} C_3-H_2\\ 2H, ABq\\ J=13 \text{ Hz} \end{array}$	R	
16	CH <sub>2</sub> OCOCH <sub>3</sub>	1780	3.88	9.4	5.83	5.15	3.5	4.67 5.04	2.00 (3H, s)	
17	CH <sub>2</sub> OCONH <sub>2</sub>	1780	3.88	10.0	5.92	5.15	3.44 <sup>a</sup> ) 3.60	4.62 4.88		
18	H <sub>2</sub> CS K <sub>S</sub>	1780	3.83	9.57	5.82	5.13	3.67 <sup>a</sup> ) 3.73	4.30 4.55		
19	H <sub>2</sub> CS L <sub>S</sub> CH <sub>2</sub> NH <sub>2</sub>	1770	3.88	9.5	5.75	5.05	3.53	4.35 <sup>b)</sup>	4.35 (2H, br)	
20	H <sub>2</sub> CS H <sub>2</sub> CS I CH <sub>2</sub> COOH	1780	3.98	9.50	5.88	5.18	3.67 <sup>a)</sup> 3.80	4.27 4.50	5.33 (2H, s)	
21	н	1785	4.06	10.0	5.85	5.18	3.64		6.52 (1H, t)	
22	$\mathbf{CH}_3$	1780	4.11	9.55	5.77	5.18	3.35 <sup>a</sup> ) 3.70		2.07 (3H, s)	
23	C1	1780	4.13	10.07	5.88	5.37	3.57 <sup>a)</sup> 4.13			

<sup>a)</sup> ABq, J=18 Hz, <sup>b)</sup> bs.

mp  $166 \sim 168^{\circ}$ C (dec), which was deprotected with methanolic hydrochloric acid in the same manner as General Procedure A (2) to give compound **20**.

Compound 19 was obtained in the same manner as the preparation of 20.

C) Synthesis of Cephalosporins **21**, **23** (*Z* isomer): A mixture of *N*,*N*-dimethylformamide (3 ml) and phosphoryl chloride (3 mmol) was stirred at 37 to 40°C for 30 minutes. To the solution were added methylene chloride (3 ml) and 2-(2-formamidopyridin-6-yl)-2-methoxyiminoacetic acid (3 mmol) at -20 to  $-25^{\circ}$ C and stirred at -10 to  $-15^{\circ}$ C for 1 hour. A solution of *p*-nitrobenzyl 7-amino-3-cephem-4-carboxylate (3 mmol) and trimethylsilylacetamide (2 g) in methylene chloride (200 ml) was added to the above solution at -10 to  $-15^{\circ}$ C, and then stirred at that temperature for 30 minutes. After the solution was concentrated, ethyl acetate and water were added to the residue. The ethyl acetate layer was separated, washed with water and then concentrated. The residue was triturated with diethyl ether to give *p*-nitrobenzyl 7-[2-(2-formamidopyridin-4-yl)-2-methoxyiminoacetamido]-3-cephem-4-carboxylate (730 mg, 45.1 %); mp 195~200°C (dec); IR (Nujol) 3350, 3200, 1790, 1725, 1690, 1660 cm<sup>-1</sup>.

10% Palladium on carbon (216 mg) was added to a solution of the above product (1 mmol) in tetrahydrofuran (10 ml), methanol (5 ml), acetic acid (0.075 ml) and water (0.75 ml). The mixture was subjected to catalytic hydrogenation at room temperature under ordinary pressure for 5 hours. After filtering off the catalyst, the filtrate was concentrated. Ethyl acetate and aqueous sodium bicarbonate were added to the residue, and the aqueous layer was separated. The aqueous solution was adjusted to pH 2 with 10% hydrochloric acid. The precipitates were collected by filtration and washed with water to give 7-[2-(2-formamidopyridin-6-yl)-2-methoxyiminoacetamido]-3-cephem-4-carboxylic acid (300 mg, 71.9%); mp 202 ~ 204°C (dec); IR (Nujol) 3250, 3200, 1780, 1720, 1660 cm<sup>-1</sup>.

A solution of the above compound (5 mmol) and conc hydrochloric acid (1 ml) in methanol (12 ml) was stirred at room temperature for 1 hour. To the resultant solution was added diethyl ether (100 ml), and the precipitates were collected by filtration and dissolved in a mixture of methanol (50 ml) and water (10 ml). The solution was adjusted to pH 3 with aqueous sodium bicarbonate, treated with activated charcoal (1 g) and concentrated to a volume of about 20 ml. The precipitating crystals were collected by filtration, washed with water and dried to give 7-[2-(2-aminopyridin-6-yl)-2-methoxyimino-acetamido]-3-cephem-4-carboxylic acid (21) (1.48 g, 76.5%); mp  $215 \sim 220^{\circ}$ C (dec).

Compound 23 was prepared from *p*-nitrobenzyl 7-amino-3-chloro-3-cephem-4-carboxylate in substantially the same manner.

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#### References

- ΤΑΚΑΥΑ, Τ.; Η. ΤΑΚΑSUGI, Τ. MASUGI, Τ. CHIBA, H. KOCHI, Τ. ΤΑΚΑΝΟ & H. NAKANO: Structure-activity relationships of sodium 7β-[(Z)-2-(2-amino-4-thiazolyl)-2-methoxyiminoacetamido]-3-cephem-4-carboxylate (ceftizoxime) and its related compounds. Nippon Kagaku Kaishi 5: 785~804, 1981
- BUCOURT, R.; R. HEYMES, A. LUTZ, L. PENASSE & J. PERRONT: Propriétés antibiotiques inattendues dans le domaine des cephalosporines. C. R. Acad. Sc. Paris, Sérié D, 284: 1847~1849, 1977
- HEYMES, R.; A. LUTZ & E. SCHRINNER: Experimental evaluation of HR 756, a new cephalosporin derivative: Pre-clinical study. Infection 5: 259 ~ 260, 1977
- OCHIAI, M.; O. AKI, A. MORIMOTO, T. OKADA & Y. MATSUSHITA: New cephalosporin derivatives with high antibacterial activities. Chem. Pharm. Bull. 25: 3115~3117, 1977
- REINER, R.; U. WEISS, U. BROMBACHER, P. LANZ, M. MONTAVON, A. FURLENMEIER, P. ANGEHRN & P. J. PROBST: Ro 13–9904/001, a novel potent and long-acting parenteral cephalosporin. J. Antibiotics 33: 783 ~ 786, 1980
- KAMIYA, T.; T. TERAJI, Y. NAKAI, K. SAKANE & J. GOTO: Cephalosporanic acid derivatives. Ger. Offen. 2,848,912, May 17, 1979 (Chem. Abst. 92: 111033g, 1980)

- OGURA, K.: Chemistry of formaldehyde dimethyl dithioacetal S-oxide (FAMSO). Yuki Gosei Kagaku Kyokaishi 37: 903~913, 1979
- 8) A) YAMADA, H.; H. TOBIKI, K. JIMPO, K. GOODA, Y. TAKEUCHI, S. UEDA, T. KOMATSU, T. OKUDA, H. NOGUCHI, K. IRIE & T. NAKAGOME: New broad-spectrum cephalosporins with anti-pseudomonal activity. II. Synthesis and antibacterial activity of 7β-[2-acylamino-2-(4-hydroxyphenyl)acetamido]-3-[(1-methyl-1*H*-tetrazol-5-yl)thiomethyl]ceph-3-em-4-carboxylic acids. J. Antibiotics 36: 532~542, 1983
  b) NAKAO, H.; H. YANAGISAWA, B. SHIMIZU, M. KANEKO, M. NAGANO & S. SUGAWARA: A new semisynthetic 7α-methoxycephalosporin, CS-1170: 7β-[[(Cyanomethyl)thio]acetamido]-7α-methoxy-3-[[(1-methyl-1*H*-tetrazol-5-yl)thio]methyl]-3-cephem-4-carboxylic acid. J. Antibiotics 29: 554~558, 1976
  c) IWANAMI, M.; T. MAEDA, M. FUJIMOTO, Y. NAGANO, N. NAGANO, A. YAMAZAKI & T. SHIBANUMA: Synthesis of new cephamycin derivatives and a novel rearrangement between isothiazolethioacetamides and 1,3-dithiethancarboxamides. Chem. Pharm. Bull. 28: 2629~2636, 1980
- 9) NUMATA, M.; I. MINAMIDA, S. TSUSHIMA, T. NISHIDA, M. YAMAOKA & N. MATSUMOTO: Synthesis of new cephalosporins with potent antibacterial activities. Chem. Pharm. Bull. 25: 3117~3119, 1977
- FERRARI, G. & B. MARCON: Pyridine series. III. 6-Aminopicolinic acid. Farmaco (Pavia) Ed. Sci. 14: 594~597, 1959 (Chem. Abst. 54: 6709a, 1960)
- FERRARI, G.: Pyridine series. I. Derivatives of 6-aminonicotinic acid. Boll. Chim. Farm. 96: 542~546, 1957 (Chem. Abst. 52: 7313g, 1958)
- FERRARI, G. & E. MARCON: Pyridine series. II. Derivatives of 2-amino-isonicotinic acid. Farmaco (Pavia) Ed. Sci. 13: 485~489, 1958 (Chem. Abst. 53: 7162c, 1959)
- 13) TADEUSZ, T. & P. EDWIN: The reaction with nitrous acid of certain derivatives of 4-aminopyridines substituted in position 2 or 2 and 6. V. Rocziki. Chem. 35: 463~474, 1961 (Chem. Abst. 55: 25943i, 1961)